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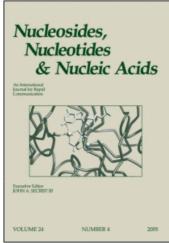
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## Nucleosides, Nucleotides and Nucleic Acids

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# Novel <i>S</i>-Ribosylhomocysteine Analogues as Potential Inhibitors of LuxS Enzyme

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# NOVEL S-RIBOSYLHOMOCYSTEINE ANALOGUES AS POTENTIAL INHIBITORS OF LUXS ENZYME

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 $\Box$  Selective cross-coupling of the protected 6-fluoro-6-iodo-α-D-ribo-hex-5-enofuranose with 2 equivalents of 4-ethoxy-4-oxobutylzinc bromide in the presence of  $Pd[P(Ph)_3]_4$  followed by deprotections gave methyl 5,6,7,8,9-pentadeoxy-6-fluoro-α/β-D-ribo-dec-5(Z)-enofuranuronate; a S-ribosylhomocysteine analogue with the sulfur and carbon-5 atoms replaced by the fluoro(vinyl) unit.

**Keywords** S-Ribosylhomocysteine; Negishi cross-coupling; vinyl fluorides; luxS enzyme

### INTRODUCTION

Hydrolytic cleavage of the S-adenosyl-L-homocysteine (SAH) by SAH hydrolase produced adenosine (Ado) and L-homocysteine (Hcy).<sup>[1]</sup> The cellular levels of SAH are critical because SAH is a potent feedback inhibitor of crucial transmethylation enzymes. Alternatively, hydrolysis of SAH by nucleosidase pfs yields adenine and S-ribosyl-L-homocysteine (SRH). S-Ribosylhomocysteinase (LuxS) enzyme catalyzes the cleavage of the thioether linkage in SRH to produce Hcy and 4,5-dihydroxy-2,3-pentanedione (DHPD) (Figure 1).<sup>[2]</sup> DHPD spontaneously cyclizes to **A** and complexes with borate to form a furanosyl borate diester, a small signaling

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FIGURE 1 Enzymatic conversion of SAH by Pfs nucleosidase and LuxS: Biosynthetic pathway to AI-2.

molecule called autoinducer (AI) of type 2. Autoinducers function in *interspecies* communications and mediate a quorum sensing process of cell-cell communication that bacteria use to coordinate gene expression in response to fluctuation in cell density.<sup>[3]</sup>

LuxS is a metalloenzyme containing an Fe<sup>2+</sup> ion coordinated by His-54, His-58, and Cys-126 and a water molecule. The native enzyme is unstable under aerobic conditions however substitution of Co<sup>2+</sup> for the native metal ion produces a highly stable variant with wild-type catalytic activity. In the proposed catalytic mechanism of LuxS, the metal ion acts as a Lewis acid, facilitating two consecutive aldose-ketose (C1  $\rightarrow$  C2) and ketose-ketose (C2  $\rightarrow$  C3) isomerization steps and a final  $\beta$ -elimination of Hcy from 3-keto intermediates.<sup>[4]</sup>

Two substrate analogues S-anhydroribosyl-L-homocysteine (**B**) and S-homoribosyl-L-cysteine (**C**) which were found to prevent the initial step and the final step of the mechanism have been recently synthesized (Figure 2).<sup>[5]</sup> Pei's laboratory prepared a series of structural analogues in which the unstable enediolate moiety formed during isomerizations was replaced with a planar hydroxamate group. The stable isostere **D** showed submicromolar inhibition of the enzyme ( $K_{\rm I} = 0.72 \ \mu {\rm M}$ ).<sup>[6]</sup> We report herein synthesis

FIGURE 2 LuxS inhibitors.

of the SRH analogues with the sulfur and carbon-5 atoms replaced by a fluoro(vinyl) unit (e.g., **6**), which are not capable to undergo the final elimination step. These ribosyl (depurinated) analogues of SAH were also designed as probes to evaluate similarities between SAH hydrolase and SRHase (LuxS).

#### RESULTS AND DISCUSSION

Treatment of the diacetone 3-O-benzovlallose with periodic acid effected the regioselective removal of 5,6-O-isopropylidene group, and sequential oxidative cleavage of the exposed vicinal diol<sup>[7]</sup> gave the ribose 5aldehyde 1 (Scheme 1). Wittig-Horner treatment of 1 with sulfonyl-stabilized fluorophosphonate<sup>[8]</sup> gave 6-(fluoro)homovinyl sulphones 2. Stannyldesulfonylation followed by iododestannylation<sup>[9]</sup> of the resulting vinyl stannanes 3 afforded the protected 5,6-dideoxy-6-fluoro-6-iodo-α-D-ribo-hex-5enofuranose 4 (E/Z, 3:2). Trans selective Negishi cross-coupling<sup>[10]</sup> of 4 using 2 equivalents of 4-ethoxy-4-oxobutylzinc bromide gave fluoro(vinyl) SRH analogue 5 (Z, 54%; 90% based on the conversion of the E isomer only). [11] Treatment of 5 with NH<sub>3</sub>/MeOH affected debenzoylation and transesterification and subsequent deacetonization with aqueous trifluoroacetic acid (TFA) gave methyl 5,6,7,8,9-pentadeoxy-6-fluoro-α/β-D-ribodec-5(Z)-enofuranuronate 6 (40%;  $\alpha/\beta$ , 3:7). To investigate whether the stereochemistry of the 3-OH group in the SRH analogues can effect a second enolization step, [4a] the xylo analogue of 6 was prepared similarly from diacetone 3-O-benzoylglucose. Moreover, debenzoylation of 3 (E/Z, 7:3) with NH<sub>3</sub>/MeOH and subsequent concomitant protiodestannylation and deacetonization with TFA/H<sub>2</sub>O afforded 5,6-dideoxy-6-fluoro-D-ribo-hex-5enofuranose (E/Z,  $\sim 3:1$ ;  $\alpha/\beta \sim 1:6$ ).

PhSO<sub>2</sub>CHFPO(OEt)<sub>2</sub>
LHMDS/THF/-78 °C

Bu<sub>3</sub>SnH/AIBN

$$\begin{array}{c} 2 \times PhSO_2 \\ 3 \times Bu_3Sn \\ \end{array}$$

BrZn(CH<sub>2</sub>)<sub>3</sub>CO<sub>2</sub>Et
Pd(PPh<sub>3</sub>)<sub>4</sub>/benzene/ $\Delta$ 

1. NH<sub>3</sub>/MeOH
2. TFA/H<sub>2</sub>O

 $\begin{array}{c} 1 \times PhSO_2 \\ \end{array}$ 
 $\begin{array}{c} 4 \times PhSO_2 \\ \end{array}$ 

SCHEME 1

The 5,6,7,8,9-pentadeoxy-6-fluoro-D-*ribo*-dec-5(*Z*)-enofuranuronate **6** and 5,6-dideoxy-6-fluoro-D-*ribo*-hex-5-enofuranose and their *xylo* epimers were evaluated as potential inhibitors of *Bacillus subtilis S*-ribosylhomocysteinase (LuxS) using the inhibition assays as described previuosly. <sup>[6]</sup> None of the compounds showed significant activity. These results might indicate that LuxS has more rigid requirements for binding than SAH hydrolase and that the intact Hcy unit in substrate/inhibitor is required for proper binding.

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- 11. Typical Procedure: Pd[P(Ph)<sub>3</sub>]<sub>4</sub> (22 mg, 0.01 mmol) was added to a stirred solution of **4** (42 mg, 0.097 mmol; E/Z, 3:2) in anhydrous benzene (4 mL) under N<sub>2</sub> at ambient temperature. After 2 minutes, 4-ethoxy-4-oxobutylzinc bromide (0.5M/THF; 0.30 mmol, 0.60 mL) was added and the resulting mixture was heated at 55°C for 5 hours. EtOAc (30 mL) and NaHCO<sub>3</sub>/H<sub>2</sub>O (10 mL) were added and the separated organic layer was washed with H<sub>2</sub>O (10 mL), NaCl/H<sub>2</sub>O (10 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), and then was evaporated. Column chromatography (10 → 30% EtOAc/hexanes) gave **5**(Z) (22 mg, 54%; 90% based on the conversion of E-isomer): <sup>1</sup>H NMR δ 1.24 (t, J = 7.1 Hz, 3,

CH<sub>3</sub>), 1.37 & 1.60 (2 × s, 2 × 3, 2 × CH<sub>3</sub>), 1.84 ("quint",  $f_{8-7/7'/9/9'} = 7.4$  Hz, 2, H8/8'), 2.25 (dt,  $f_{7\text{-}F} = 17.6$  Hz,  $f_{7-8/8'} = 7.5$  Hz, 2, H7/7'), 2.32 (t,  $f_{9-8/8'} = 7.4$  Hz, 2, H9/9'), 4.09 (q,  $f_{8-7} = 7.1$  Hz, 2, CH<sub>2</sub>), 4.72 (dd,  $f_{8-4} = 9.2$  Hz,  $f_{3-2} = 4.7$  Hz, 1, H3), 4.75 (dd,  $f_{5-F} = 35.0$  Hz,  $f_{5-4} = 8.9$  Hz, 1, H5), 4.95 (t,  $f_{2-1/3} = 4.3$  Hz, 1, H2), 5.19 (t,  $f_{4-3/5} = 9.1$  Hz, 1, H4), 5.89 (d,  $f_{1-2} = 3.8$  Hz, 1, H1), 7.48–8.09 (m, 5, Ar); <sup>19</sup>F NMR  $\delta$  –102.14 (dt,  $f_{F-H5} = 34.1$  Hz,  $f_{F-H7/7'} = 17.6$  Hz); HRMS (AP-ESI)  $f_{7-1} = 17.6$  Hz (MH<sup>+</sup>) 423.1814; found 423.1815.